

RSPT 1050 - Module G: Oxygen Transport

XIII. Equations

Given the following, calculate the CaO₂, CvO₂ and the Ca-vDO₂:

- A. Hb: 6 gm% SaO₂: 84% PaO₂: 80 mm Hg SvO₂: 66% PvO₂: 43 mm Hg
 $\text{CaO}_2: (\text{Hb} \times 1.34 \times \text{SaO}_2) + (\text{PaO}_2 \times .003) = (6 \times 1.34 \times .84) + (80 \times .003) = 6.75 + .24 = 7.0 \text{ vol\%}$
 $\text{CvO}_2: (\text{Hb} \times 1.34 \times \text{SvO}_2) + (\text{PvO}_2 \times .003) = (6 \times 1.34 \times .66) + (43 \times .003) = 5.31 + .13 = 5.4 \text{ vol\%}$
 $\text{C}_{(\text{a-v})}\text{O}_2: \text{CaO}_2 - \text{CvO}_2 = 7.0 - 5.4 = 1.6 \text{ vol\%}$
- B. Hb: 12 gm% SaO₂: 67% PaO₂: 55 mm Hg SvO₂: 45% PvO₂: 35 mm Hg
 $\text{CaO}_2: (\text{Hb} \times 1.34 \times \text{SaO}_2) + (\text{PaO}_2 \times .003) = (12 \times 1.34 \times .67) + (55 \times .003) = 10.77 + .17 = 10.9 \text{ vol\%}$
 $\text{CvO}_2: (\text{Hb} \times 1.34 \times \text{SvO}_2) + (\text{PvO}_2 \times .003) = (12 \times 1.34 \times .45) + (35 \times .003) = 7.24 + .11 = 7.4 \text{ vol\%}$
 $\text{C}_{(\text{a-v})}\text{O}_2: \text{CaO}_2 - \text{CvO}_2 = 10.9 - 7.4 = 3.5 \text{ vol\%}$
- C. Hb: 14.5 gm% SaO₂: 88% PaO₂: 78 mm Hg SvO₂: 70% PvO₂: 40 mm Hg
 $\text{CaO}_2: (\text{Hb} \times 1.34 \times \text{SaO}_2) + (\text{PaO}_2 \times .003) = (14.5 \times 1.34 \times .88) + (78 \times .003) = 17.10 + .23 = 17.3 \text{ vol\%}$
 $\text{CvO}_2: (\text{Hb} \times 1.34 \times \text{SvO}_2) + (\text{PvO}_2 \times .003) = (14.5 \times 1.34 \times .7) + (40 \times .003) = 13.60 + .12 = 13.7 \text{ vol\%}$
 $\text{C}_{(\text{a-v})}\text{O}_2: \text{CaO}_2 - \text{CvO}_2 = 17.3 - 13.7 = 3.6 \text{ vol\%}$
- D. Hb: 8.8 gm% SaO₂: 75% PaO₂: 40 mm Hg SvO₂: 66% PvO₂: 35 mm Hg
 $\text{CaO}_2: (\text{Hb} \times 1.34 \times \text{SaO}_2) + (\text{PaO}_2 \times .003) = (8.8 \times 1.34 \times .75) + (40 \times .003) = 8.84 + .12 = 9.0 \text{ vol\%}$
 $\text{CvO}_2: (\text{Hb} \times 1.34 \times \text{SvO}_2) + (\text{PvO}_2 \times .003) = (8.8 \times 1.34 \times .66) + (35 \times .003) = 7.78 + .11 = 7.9 \text{ vol\%}$
 $\text{C}_{(\text{a-v})}\text{O}_2: \text{CaO}_2 - \text{CvO}_2 = 9.0 - 7.9 = 1.1 \text{ vol\%}$
- E. Hb: 10 gm% SaO₂: 78% PaO₂: 55 mm Hg SvO₂: 69% PvO₂: 36 mm Hg
 $\text{CaO}_2: (\text{Hb} \times 1.34 \times \text{SaO}_2) + (\text{PaO}_2 \times .003) = (10 \times 1.34 \times .78) + (55 \times .003) = 10.45 + .17 = 10.6 \text{ vol\%}$
 $\text{CvO}_2: (\text{Hb} \times 1.34 \times \text{SvO}_2) + (\text{PvO}_2 \times .003) = (10 \times 1.34 \times .69) + (36 \times .003) = 9.25 + .11 = 9.4 \text{ vol\%}$
 $\text{C}_{(\text{a-v})}\text{O}_2: \text{CaO}_2 - \text{CvO}_2 = 10.6 - 9.4 = 1.2 \text{ vol\%}$

**ITEMS F THROUGH J ARE IMPOSSIBLE. PaO₂ CAN NEVER BE LESS THAN PvO₂.
TRY DOING THE PROBLEMS WITH THE PaO₂ & SaO₂ REVERSED WITH THE VENOUS VALUES**

- F. Hb: 10 gm% SvO₂: 75% PvO₂: 40 mm Hg SaO₂: 87% PaO₂: 77 mm Hg
 $\text{CaO}_2: (\text{Hb} \times 1.34 \times \text{SaO}_2) + (\text{PaO}_2 \times .003) = (10 \times 1.34 \times .87) + (77 \times .003) = 11.66 + .23 = 11.9 \text{ vol\%}$
 $\text{CvO}_2: (\text{Hb} \times 1.34 \times \text{SvO}_2) + (\text{PvO}_2 \times .003) = (10 \times 1.34 \times .75) + (40 \times .003) = 10.05 + .12 = 10.2 \text{ vol\%}$
 $\text{C}_{(\text{a-v})}\text{O}_2: \text{CaO}_2 - \text{CvO}_2 = 11.9 - 10.2 = 1.7 \text{ vol\%}$
- G. Hb: 12 gm% SvO₂: 66% PvO₂: 44 mm Hg SaO₂: 93% PaO₂: 89 mm Hg
 $\text{CaO}_2: (\text{Hb} \times 1.34 \times \text{SaO}_2) + (\text{PaO}_2 \times .003) = (12 \times 1.34 \times .93) + (89 \times .003) = 14.95 + .27 = 15.2 \text{ vol\%}$
 $\text{CvO}_2: (\text{Hb} \times 1.34 \times \text{SvO}_2) + (\text{PvO}_2 \times .003) = (12 \times 1.34 \times .66) + (44 \times .003) = 10.61 + .13 = 10.7 \text{ vol\%}$
 $\text{C}_{(\text{a-v})}\text{O}_2: \text{CaO}_2 - \text{CvO}_2 = 15.2 - 10.7 = 4.5 \text{ vol\%}$
- H. Hb: 8.5 gm% SvO₂: 68% PvO₂: 55 mm Hg SaO₂: 95% PaO₂: 83 mm Hg
 $\text{CaO}_2: (\text{Hb} \times 1.34 \times \text{SaO}_2) + (\text{PaO}_2 \times .003) = (8.5 \times 1.34 \times .95) + (83 \times .003) = 10.82 + .25 = 11.1 \text{ vol\%}$
 $\text{CvO}_2: (\text{Hb} \times 1.34 \times \text{SvO}_2) + (\text{PvO}_2 \times .003) = (8.5 \times 1.34 \times .68) + (55 \times .003) = 7.75 + .19 = 7.9 \text{ vol\%}$
 $\text{C}_{(\text{a-v})}\text{O}_2: \text{CaO}_2 - \text{CvO}_2 = 11.1 - 7.9 = 3.2 \text{ vol\%}$
- I. Hb: 14 gm% SvO₂: 75% PvO₂: 40 mm Hg SaO₂: 79% PaO₂: 60 mm Hg
 $\text{CaO}_2: (\text{Hb} \times 1.34 \times \text{SaO}_2) + (\text{PaO}_2 \times .003) = (14 \times 1.34 \times .79) + (60 \times .003) = 14.82 + .18 = 15.0 \text{ vol\%}$
 $\text{CvO}_2: (\text{Hb} \times 1.34 \times \text{SvO}_2) + (\text{PvO}_2 \times .003) = (14 \times 1.34 \times .75) + (40 \times .003) = 14.07 + .12 = 14.2 \text{ vol\%}$
 $\text{C}_{(\text{a-v})}\text{O}_2: \text{CaO}_2 - \text{CvO}_2 = 15.0 - 14.2 = 0.8 \text{ vol\%}$
- J. Hb: 18 gm% SvO₂: 78% PvO₂: 50 mm Hg SaO₂: 96% PaO₂: 93 mm Hg
 $\text{CaO}_2: (\text{Hb} \times 1.34 \times \text{SaO}_2) + (\text{PaO}_2 \times .003) = (18 \times 1.34 \times .96) + (93 \times .003) = 23.12 + .28 = 23.4 \text{ vol\%}$
 $\text{CvO}_2: (\text{Hb} \times 1.34 \times \text{SvO}_2) + (\text{PvO}_2 \times .003) = (18 \times 1.34 \times .78) + (50 \times .003) = 18.81 + .15 = 19.0 \text{ vol\%}$
 $\text{C}_{(\text{a-v})}\text{O}_2: \text{CaO}_2 - \text{CvO}_2 = 23.4 - 19.0 = 4.4 \text{ vol\%}$

Given the following information, calculate the % shunt

Hb: 13g% P_{baro} : 750 mm Hg PaO_2 : 50 mm Hg FiO_2 : 70% PaCO_2 : 43 mm Hg
 SaO_2 : 85% PvO_2 : 37 mm Hg SvO_2 : 65%

- K. $\text{PAO}_2 = [(P_{\text{baro}} - P_{\text{H}_2\text{O}}) \times \text{FiO}_2] - (\text{PaCO}_2 \times 1.25) = [(750 - 47) \times .7] - (43 \times 1.25) = (703 \times .7) - 53.75 = 492.1 - 53.75 = 438 \text{ mm Hg}$
- L. $\text{CcO}_2 = (\text{Hb} \times 1.34 \times 1.0) + (\text{PAO}_2 \times .003) = (13 \times 1.34) + (438 \times .003) = 17.42 + 1.31 = 18.73 \text{ vol\%}$
- M. $\text{CaO}_2 = (\text{Hb} \times 1.34 \times \text{SaO}_2) + (\text{PaO}_2 \times .003) = (13 \times 1.34 \times .85) + (50 \times .003) = 14.81 + 0.15 = 14.96 \text{ vol\%}$
- N. $\text{CvO}_2 = (\text{Hb} \times 1.34 \times \text{SvO}_2) + (\text{PvO}_2 \times .003) = (13 \times 1.34 \times .65) + (37 \times .003) = 11.32 + 0.11 = 11.43 \text{ vol\%}$
- O.
$$\frac{Q_s}{Q_t} = \frac{(Cc' O_2 - CaO_2)}{(Cc' O_2 - CvO_2)} = \frac{(18.73 - 14.96)}{(18.73 - 11.43)} = \frac{3.77}{7.3} = .516 = 51.6\%$$

Given the following information, calculate the % shunt

Hb: 8g% P_{baro} : 740 mm Hg PaO_2 : 67 mm Hg FiO_2 : 70% PaCO_2 : 55 mm Hg
 SaO_2 : 92% PvO_2 : 38 mm Hg SvO_2 : 68%

- P. $\text{PAO}_2 = [(P_{\text{baro}} - P_{\text{H}_2\text{O}}) \times \text{FiO}_2] - (\text{PaCO}_2 \times 1.25) = [(740 - 47) \times .7] - (55 \times 1.25) = (693 \times .7) - 68.75 = 485.1 - 68.75 = 416 \text{ mm Hg}$
- Q. $\text{CcO}_2 = (\text{Hb} \times 1.34 \times 1.0) + (\text{PAO}_2 \times .003) = (8 \times 1.34) + (416 \times .003) = 10.72 + 1.25 = 11.97 \text{ vol\%}$
- R. $\text{CaO}_2 = (\text{Hb} \times 1.34 \times \text{SaO}_2) + (\text{PaO}_2 \times .003) = (8 \times 1.34 \times .92) + (67 \times .003) = 9.86 + .20 = 10.06 \text{ vol\%}$
- S. $\text{CvO}_2 = (\text{Hb} \times 1.34 \times \text{SvO}_2) + (\text{PvO}_2 \times .003) = (8 \times 1.34 \times .68) + (38 \times .003) = 7.29 + 0.11 = 7.4 \text{ vol\%}$
- T.
$$\frac{Q_s}{Q_t} = \frac{(Cc' O_2 - CaO_2)}{(Cc' O_2 - CvO_2)} = \frac{(11.97 - 10.06)}{(11.97 - 7.4)} = \frac{1.91}{4.57} = .418 = 41.8\%$$